

DECLARATION

I, Naoto Noda of Nishikawa & Associates, Visual City, Suite 401, 43-9, Higashi-Nippori 3-chome, Arakawa-ku, Tokyo, Japan, do solemnly and sincerely declare that I am well acquainted with both the Japanese language and the English language and that the attached English translation of an officially certified copy of Patent Application No. 2003-123004 is a true and correct translation to the best of my knowledge and belief from the Japanese language to the English language.

Dated this 14th day of October, 2009

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(translator)

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JAPAN PATENT OFFICE

This is to certify that the annexed is a true copy of
the following application as filed with this Office.

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[List of Materials Submitted]

[Material Name]	Specification	1
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[Material Name]	Abstract	1
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[Document Name] Specification

[Title of the Invention] Heat-Shrinkable Layered Film and
Package Made with the Same

[Claims]

5 **[Claim 1]** A heat-shrinkable multi-layer film comprising a
heat-shrinkable support film (base film) and, on at least one
surface of the base film, at least one layer structure
including a layer (a) formed of a poly(carboxylic acid)
polymer (A) and a layer (b) formed of a polyvalent metal
10 compound (B), the layers (a) and (b) being in contact with
each other, and the multi-layer film exhibiting a percent
thermal shrinkage of 5 to 90%.

[Claim 2] A heat-shrinkable multi-layer film according to
claim 1, wherein the heat-shrinkable support film (base film)
15 exhibits a percent thermal shrinkage of 5 to 90%.

[Claim 3] A heat-shrinkable multi-layer film according to
claim 1 or 2, which exhibits an oxygen permeability of 500
 $\text{cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less as measured at 30°C and a relative
humidity of 80%.

20 **[Claim 4]** A heat-shrinkable multi-layer film according to
any one of claims 1 to 3, wherein the polyvalent metal
compound (B) is a divalent metal compound.

[Claim 5] A heat-shrinkable multi-layer film according to
any one of claims 1 to 4, wherein the poly(carboxylic acid)
25 polymer (A) is a homopolymer or copolymer formed of at least
one polymerizable monomer selected from among acrylic acid,
maleic acid, and methacrylic acid, and/or a mixture of such

homopolymers or copolymers.

[Claim 6] A heat-shrinkable multi-layer film according to any one of claims 1 to 5, which contains an additional layer.

[Claim 7] A heat-shrinkable multi-layer film according to
5 claim 6, wherein the additional layer is an adhesive-
containing layer.

[Claim 8] A heat-shrinkable multi-layer film according to any one of claims 1 to 7, which, after thermal shrinkage, exhibits an oxygen permeability equal to or lower than that
10 before thermal shrinkage.

[Claim 9] A heat-shrinkable multi-layer film according to any one of claims 1 to 8, which is in the form of a film, a bag, a sheet, a label, or a container.

[Claim 10] A packaged material obtained by packaging an
15 object with a heat-shrinkable multi-layer film as recited in any one of claims 1 to 9 and subjecting the thus-packaged object to thermal shrinkage treatment, wherein the heat-shrinkable multi-layer film exhibits an oxygen permeability of $500 \text{ cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less as measured at 30°C and a
20 relative humidity of 80% after the thermal shrinkage treatment.

[Detailed Description of the Invention]

[0001]

[Technical Field to which the Invention Belongs]

25 The present invention relates to a heat-shrinkable multi-layer film, and to a packaged product obtained by packaging an object with the film and subjecting the thus-

packaged object to thermal shrinkage treatment. More specifically, the present invention relates to a heat-shrinkable multi-layer film comprising a heat-shrinkable support film (base film) and, on at least one surface of the base film, at least one layer structure including a layer formed of a poly(carboxylic acid) polymer and a layer formed of a polyvalent metal compound, the layers being in contact with each other, and a packaged material obtained by packaging an object with the heat-shrinkable multi-layer film.

10 [0002]

 [Prior Art]

 In a conventional hermetic container requiring oxygen-barrier property, a container member and a cover member are formed of a material exhibiting oxygen-barrier property. In some cases, such a hermetic container *per se* is protected with a stretchable packaging film or a stretchable/shrinkable packaging film. In the case of packaging of an object without using a container member and a cover member, the object is packaged directly with a film exhibiting oxygen-barrier property and stretchability/shrinkability. As such a packaging method, for example, Patent Art. 1 describes a packaging method with a barrier·stretch·shrink film, in which after the barrier·stretch·shrink film is formed into a cylinder, and an object is inserted into the cylinder in such a manner that a film joint of the cylinder is located at the bottom of the object, the cylindrical film is cut at both front and rear ends of the object, and at the same time the

film ends are folded back to the bottom of the object and fusion-bonded to each other, and moreover the films are fusion-bonded to each other at the bottom of the object by a hot plate and then shrunk in a shrink tunnel. Patent Art. 2
5 discloses, as a film exhibiting oxygen-barrier property and shrinkability, a barrier shrink film which is formed of an aliphatic polyamide and a xylylene-containing polyamide and exhibits specific percent shrinkage and oxygen permeability under specific conditions.

10 Thus, such a film exhibiting shrinkability and gas-barrier property has conventionally been employed for packaging, and packaging films exhibiting various characteristics have been proposed.

[0003]

15 [Patent Art. 1]

Japanese Patent Application Laid-Open (*kokai*) No. 6-179466

[Patent Art. 2]

Japanese Patent Application Laid-Open (*kokai*) No. 2001-341201

[0004]

20 **[Problems Sought for Solution by the Invention]**

An object of the present invention is to provide a heat-shrinkable multi-layer film comprising a layer formed of a poly(carboxylic acid) polymer; a layer formed of a polyvalent metal compound; and a heat-shrinkable support film.

25 **[0005]**

[Means for solution of the Problems]

The present inventors have conducted studies on a gas-

barrier multi-layer film including a layer formed of a poly(carboxylic acid) polymer and a layer formed of a polyvalent metal compound (Japanese Patent Application No. 2002-121246), and have found that when the gas-barrier multi-layer film is formed on a heat-shrinkable base film, the thus-formed multi-layer film exhibits heat shrinkability without impairing gas-barrier property. The present invention has been accomplished on the basis of this finding.

[0006]

Accordingly, the present invention provides a heat-shrinkable multi-layer film comprising a heat-shrinkable support film (base film) and, on at least one surface of the base film, at least one layer structure including a layer (a) formed of a poly(carboxylic acid) polymer (A) and a layer (b) formed of a polyvalent metal compound (B), the layers (a) and (b) being in contact with each other, and the multi-layer film exhibiting a percent thermal shrinkage of 5 to 90%. The present invention also provides the heat-shrinkable multi-layer film of the aforementioned invention wherein the heat-shrinkable support film (base film) exhibits a percent thermal shrinkage of 5 to 90%.

Also the present invention provides the heat-shrinkable multi-layer film of the aforementioned invention which exhibits an oxygen permeability of $500 \text{ cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less as measured at 30°C and a relative humidity of 80%. The present invention provides the heat-shrinkable multi-layer film of the aforementioned invention wherein the polyvalent

metal compound (B) is a divalent metal compound. The present invention provides the heat-shrinkable multi-layer film of the aforementioned invention wherein the poly(carboxylic acid) polymer (A) is a homopolymer or copolymer formed of at least one polymerizable monomer selected from among acrylic acid, maleic acid, and methacrylic acid, and/or a mixture of such homopolymers or copolymers. The present invention provides the heat-shrinkable multi-layer film of the aforementioned invention which contains an additional layer.

10 The present invention provides the heat-shrinkable multi-layer film of the aforementioned invention wherein the additional layer is an adhesive-containing layer. The present invention also provides the heat-shrinkable multi-layer film of the aforementioned invention which, after

15 thermal shrinkage, exhibits an oxygen permeability equal to or lower than that before thermal shrinkage.

The present invention further provides the heat-shrinkable multi-layer film of the aforementioned invention which is in the form of a film, a bag, a sheet, a label, or a

20 container. The present invention still further provides a packaged product obtained by packaging an object with the heat-shrinkable multi-layer film of the aforementioned invention and subjecting the thus-packaged object to thermal shrinkage treatment.

25 [0007]

[Mode for Carrying Out the Invention]

A heat-shrinkable multi-layer film according to the

present invention (hereinafter may be abbreviated as "the multi-layer film") includes a heat-shrinkable support film (base film) and, on at least one surface of the base film, at least one layer structure including a layer (a) formed of a poly(carboxylic acid) polymer (A) and a layer (b) formed of a polyvalent metal compound (B), the layers (a) and (b) being in contact with each other. Shrinkage of the entirety of the heat-shrinkable multi-layer film is determined by the shrinkage of the heat-shrinkable support film (base film).

10 [0008]

The polymer (A) making up the layer (a) formed of the poly(carboxylic acid) polymer (A) employed in the present invention may be an existing poly(carboxylic acid) polymer. The term "existing poly(carboxylic acid) polymer" refers a polymer containing in the molecule thereof two or more carboxyl groups. Specific examples of the existing poly(carboxylic acid) polymer include homopolymers or copolymers formed from an α,β -monoethylenic unsaturated carboxylic acid as a polymerizable monomer; copolymers formed from an α,β -monoethylenic unsaturated carboxylic acid and an ethylenic unsaturated monomer; and acidic polysaccharides containing in the molecule a carboxyl group, such as alginic acid and pectin. These poly(carboxylic acid) polymers (A) may be employed singly or in combination of two or more species.

Typical examples of the α,β -monoethylenic unsaturated carboxylic acid include acrylic acid, methacrylic acid,

itaconic acid, maleic acid, fumaric acid, and crotonic acid. Typical examples of the ethylenic unsaturated monomer capable of being copolymerized with such an α,β -monoethylenic unsaturated carboxylic acid include olefins such as ethylene and propylene; saturated carboxylic acid vinyl esters such as vinyl acetate; alkyl acrylates; alkyl methacrylates; alkyl itaconates; acrylonitrile; halogen-containing monomers such as vinyl chloride, vinylidene chloride, vinyl fluoride, and vinylidene fluoride; and aromatic vinyl monomers such as styrene. In the case where the poly(carboxylic acid) polymer (A) is a copolymer formed from an α,β -monoethylenic unsaturated carboxylic acid and a saturated carboxylic acid vinyl ester (e.g., vinyl acetate), the copolymer may be subjected to saponification before use, to thereby convert saturated carboxylic acid vinyl ester moieties into vinyl alcohol.

[0009]

In the case where the poly(carboxylic acid) polymer (A) is a copolymer formed from an α,β -monoethylenic unsaturated carboxylic acid and an ethylenic unsaturated monomer, from the viewpoints of gas-barrier property, high-temperature water vapor resistance, and hot water resistance of the multi-layer film of the present invention, the compositional proportion of the α,β -monoethylenic unsaturated carboxylic acid in the copolymer is preferably 60 mol% or more, more preferably 80 mol% or more, much more preferably 90 mol% or more, most preferably 100 mol% (i.e., the poly(carboxylic

acid) polymer (A) is a polymer formed solely from the α,β -monoethylenic unsaturated carboxylic acid). In the case where the poly(carboxylic acid) polymer (A) is a polymer formed solely from the α,β -monoethylenic unsaturated carboxylic acid, examples of the polymer include homopolymers or copolymers formed through polymerization of at least one polymerizable monomer selected from among acrylic acid, methacrylic acid, itaconic acid, maleic acid, fumaric acid and crotonic acid, and/or mixtures of such homopolymers or copolymers. More preferably, there are employed polyacrylic acid, polymethacrylic acid, polymaleic acid, and mixtures thereof. In the case where the poly(carboxylic acid) polymer (A) is a substance other than a polymer formed from an α,β -monoethylenic unsaturated carboxylic acid monomer; for example, an acidic polysaccharide, alginic acid is preferably employed.

So long as properties such as a gas-barrier property and heat shrinkability of the multi-layer film are not impaired, the entirety or a portion of carboxyl groups contained in the molecule of the poly(carboxylic acid) polymer (A) may be formed into a metal salt by use of a compound of a monovalent metal (e.g., sodium or potassium), and the resultant poly(carboxylic acid) polymer may be employed singly or in combination with another poly(carboxylic acid) polymer. Alternatively, the poly(carboxylic acid) polymer (A) may be employed in combination with the aforementioned monovalent metal compound.

[0010]

No particular limitation is imposed on the number average molecular weight of the poly(carboxylic acid) polymer (A), but, from the viewpoint of film formability, the number average molecular weight is preferably 2,000 to 10,000,000, more preferably 5,000 to 1,000,000.

[0011]

A compound making up the layer (b) formed of the polyvalent metal compound (B) employed in the present invention encompasses a polyvalent metal element having a valence of 2 or more, and a compound of such a polyvalent metal element. Specific examples of the polyvalent metal include alkaline earth metals such as beryllium, magnesium, and calcium; transition metals such as titanium, zirconium, chromium, manganese, iron, cobalt, nickel, copper, and zinc; and aluminum. Specific examples of the polyvalent metal compound include oxides, hydroxides, carbonates, organic acid salts, and inorganic acid salts of the aforementioned polyvalent metals; ammonium complexes of the polyvalent metals; secondary to quaternary amine complexes of the polyvalent metals; and carbonates and organic acid salts of such complexes. Examples of the organic acid salts include acetates, oxalates, citrates, lactates, phosphates, phosphites, hypophosphites, stearates, and monoethylenic unsaturated carboxylic acid salts. Examples of the inorganic acid salts include chlorides, sulfates, and nitrates. Other examples of the polyvalent metal compound include alkyl

alkoxides of the aforementioned polyvalent metals.

[0012]

These polyvalent metal compounds may be employed singly or in combination of two or more species. Of these
5 polyvalent metal compounds, the polyvalent metal compound (B) employed in the present invention is preferably a divalent metal compound, from the viewpoints of gas-barrier property, high-temperature water vapor resistance, hot water resistance, and productivity of the multi-layer film of the present
10 invention. More preferably, the polyvalent metal compound (B) is an oxide, hydroxide, or carbonate of an alkaline earth metal, cobalt, nickel, copper, or zinc; an ammonium complex of cobalt, nickel, copper, or zinc; or a carbonate of such a complex. Much more preferably, the polyvalent metal compound
15 (B) is an oxide, hydroxide, or carbonate of magnesium, calcium, copper, or zinc; an ammonium complex of copper or zinc; or a carbonate of such a complex.

[0013]

No particular limitation is imposed on the shape of the
20 polyvalent metal compound (B). However, in the case where the polyvalent metal compound (B) to be employed is in the form of particles, preferably, the particle size is small, from the viewpoint of transparency of the multi-layer film. Furthermore, from the viewpoint of, upon preparing the below-
25 described coating mixture for producing the multi-layer film of the present invention, preparation efficiency and obtaining more uniformly coating mixture, preferably, the

polyvalent metal compound is in the form of particles having a small particle size. The average particle size of the polyvalent metal compound is preferably 5 μm or less, more preferably 1 μm or less, most preferably 0.1 μm or less.

5 [0014]

As described above, the support film according to the present invention is to be a layer which determines heat shrinkability of the entirety of the multi-layer film. No particular limitation is imposed on the type of the support film employed in the present invention, so long as the film is a plastic film having the below-described properties. Specific examples of the material of such a plastic film include chlorine-containing polymers such as polyvinyl chloride and polyvinylidene chloride, and copolymers thereof; 10 polystyrene polymers; polyester polymers such as polyethylene terephthalate and polyethylene naphthalate, and copolymers thereof; polyamide polymers and copolymers thereof, such as nylon 6, nylon 66, nylon 12, nylon 6/66 copolymers, and nylon 6/12 copolymers; vinyl-acetate-containing copolymers such as 20 ethylene-vinyl acetate copolymers and ethylene-vinyl alcohol copolymers; and olefin polymers such as low-density polyethylene, linear low-density polyethylene, and polypropylene, and copolymers thereof. The support film may be a heat-shrinkable stretched sheet or film formed of such a 25 plastic material.

[0015]

The heat-shrinkable support film (base film) determines

the shrinkability of the entirety of the heat-shrinkable multi-layer film of the present invention as described above. The percent thermal shrinkage of the support film is 5 to 90%, preferably 10 to 90%, as measured in at least one direction.

5 It is extremely difficult to produce a film having the percent thermal shrinkage exceeds 90%.

No particular limitation is imposed on gas-barrier property of the support film because gas-barrier property of the entirety of the multi-layer film is secured by means of
10 the gas-barrier multi-layer structure including the layer (a) formed of a poly(carboxylic acid) polymer (A) and the layer (b) formed of a polyvalent metal compound (B). Percent thermal shrinkage of the multi-layer film of the present invention is 5 to 90%, preferably 10 to 80%, more preferably
15 15 to 75%, as measured in at least one direction.

In the present invention, the percent thermal shrinkage of the multi-layer film or the support film is measured, for example, after the film is exposed under the environment of the temperature at 50-130°C, preferably 70-120°C, more
20 preferably 80-100°C for, for example, 1 second to 120 minutes, preferably 1 second to 60 minutes, more preferably 1 second to 30 minutes. Thermal shrinkage treatment is preferably carried out in, for example, hot water, vapor, steam, or hot air. For example, the percent thermal shrinkage of the
25 multi-layer film or the support film is measured after the film is immersed in hot water at 90°C for 30 seconds.

[0016]

Next will be described the method for producing the multi-layer film of the present invention. A coating method is employed for forming, on the heat-shrinkable support film, the layer (a) containing a poly(carboxylic acid) polymer (A) and the layer (b) containing a polyvalent metal compound (B).

As used herein, the term "coating method" is a method in which a coating liquid containing a poly(carboxylic acid) polymer (A) and a solvent, or a coating liquid containing a polyvalent metal compound (B) and a solvent is applied onto the support film, and then the solvent is removed through evaporation or a similar technique, to thereby form the layer (a) or the layer (b) on the support film. The coating method is specifically performed by use of, for example, a coater or a printing apparatus. Examples of the coater, printing apparatus, and coating technique which may be employed include coating techniques such as a direct gravure technique, a reverse gravure technique, a kiss reverse gravure technique, and an offset gravure technique; and coaters employed in these techniques, such as a gravure coater, a reverse roll coater, a micro-gravure coater, an air knife coater, a dip coater, a bar coater, a comma coater, and a die coater. The coating method encompasses a method in which a coating liquid containing a monomer of the poly(carboxylic acid) polymer (A) is applied onto the support film, and the monomer is polymerized through irradiation with UV rays or electron beams, to thereby form the layer (a); and a method in which a monomer of the polymer (A) is deposited onto the support film

while the monomer is polymerized through irradiation with, for example, electron beams, to thereby form the layer (a). Also, the coating method encompasses a method in which the polyvalent metal compound (B) is deposited onto the support film through vapor deposition, sputtering, ion plating, or a similar technique, to thereby form the layer (b) containing the compound (B).

[0017]

After a coating liquid containing a poly(carboxylic acid) polymer (A) and a solvent, or a coating liquid containing a polyvalent metal compound (B) and a solvent is applied onto the support film, the solvent is evaporated for drying. No particular limitation is imposed on the method for evaporating the solvent. Examples of the evaporation method which may be employed include a method in which the solvent is naturally evaporated, a method in which the solvent is evaporated in an oven whose temperature is regulated at a predetermined level, and a method in which the solvent is evaporated in a drying apparatus equipped with the aforementioned coater, such as an arch dryer, a floating dryer, a drum dryer, or an infrared dryer. The evaporation conditions may be arbitrarily determined, so long as the support film, the layer (a) formed of a poly(carboxylic acid) polymer (A), or the layer (b) formed of a polyvalent metal compound (B) is not damaged by heat.

[0018]

No particular limitation is imposed on the order for

applying, onto the support film, a coating liquid containing a poly(carboxylic acid) polymer (A) and a solvent, and a coating liquid containing a polyvalent metal compound (B) and a solvent, so long as at least one layer structure including the layer (a) and the layer (b) which are in contact with each other is formed. Each of the layers (a) and (b) may be provided in a plurality of numbers, and in this case, layers (a) and layers (b) may be laminated alternately, or may form a sandwich structure. No particular limitation is imposed on the total of the thicknesses of the layer(s) (a) and the layer(s) (b) formed on the support film, but the total thickness is preferably 0.002 μm to 1 mm, more preferably 0.02 μm to 100 μm , much more preferably 0.1 μm to 20 μm .

The thickness of a single layer (a) is preferably 0.001 μm to 200 μm , more preferably 0.01 μm to 50 μm , much more preferably 0.05 μm to 10 μm . The thickness of a single layer (b) is preferably 0.001 μm to 800 μm , more preferably 0.01 μm to 50 μm , much more preferably 0.05 μm to 10 μm .

[0019]

The coating liquid containing a poly(carboxylic acid) polymer (A) and a solvent can be prepared by dissolving or dispersing a polymer (A) in the solvent. No particular limitation is imposed on the solvent to be employed, so long as it can uniformly dissolve or disperse the poly(carboxylic acid) polymer (A). Specific examples of the solvent include water, acetone, methyl alcohol, ethyl alcohol, isopropyl alcohol, dimethyl sulfoxide, dimethylformamide, and

dimethylacetamide. The poly(carboxylic acid) polymer (A) may be readily reacted with the polyvalent metal compound (B) in an aqueous solution, resulting in formation of non-uniform precipitates. Therefore, in the case where the coating liquid containing a poly(carboxylic acid) polymer (A) and a solvent is applied onto the layer (b) formed of a polyvalent metal compound (B), when the solvent is water, the poly(carboxylic acid) polymer (A) may be reacted with the polyvalent metal compound during application of the coating liquid, resulting in formation of non-uniform precipitates. Therefore, the solvent employed in the coating liquid is preferably a non-aqueous solvent, or a mixture of a non-aqueous solvent and water. As used herein, the term "non-aqueous solvent" means a solvent other than water.

15 **[0020]**

No particular limitation is imposed on the amount of the poly(carboxylic acid) polymer (A) contained in the coating liquid comprising the poly(carboxylic acid) polymer (A) and the solvent. However, from the viewpoint of coating performance, the total amount is preferably 0.1 to 50 wt.%, more preferably 1 to 50 wt.%. The coating liquid may optionally contain, in addition to the poly(carboxylic acid) polymer (A) and the solvent, an additive such as a polymer other than the polymer (A) (e.g., polyvinyl alcohol), a softening agent (e.g., glycerin), a stabilizer, an anti-blocking agent, a tackifier, or an inorganic layered compound (e.g., montmorillonite), so long as such an additive does not

impede gas-barrier property of the multi-layer film of the present invention, which is a final product. The total amount of such additives to be added is preferably 5 wt.% or less, more preferably 3 wt.% or less, most preferably 1 wt.% or less, on the basis of the amount of the poly(carboxylic acid) polymer (A) contained in the coating liquid.

Similar to the case described above, the coating liquid may contain a monovalent metal compound, so long as the compound does not impede gas-barrier property of the multi-layer film of the present invention, which is a final product.

[0021]

The coating liquid containing the polyvalent metal compound (B) and a solvent can be prepared by dissolving or dispersing the compound (B) in the solvent. No particular limitation is imposed on the solvent to be employed, so long as it can uniformly dissolve or disperse the polyvalent metal compound (B). Specific examples of the solvent which may be employed include water, methyl alcohol, ethyl alcohol, isopropyl alcohol, n-propyl alcohol, n-butyl alcohol, n-pentyl alcohol, dimethyl sulfoxide, dimethylformamide, dimethylacetamide, toluene, hexane, heptane, cyclohexane, acetone, methyl ethyl ketone, diethyl ether, dioxane, tetrahydrofuran, ethyl acetate, and butyl acetate. As described above, the poly(carboxylic acid) polymer (A) may be readily reacted with the polyvalent metal compound in an aqueous solution, resulting in formation of non-uniform precipitates. Therefore, in the case where the coating

liquid containing the polyvalent metal compound (B) and a solvent is applied onto the layer (a) formed of a poly(carboxylic acid) polymer (A), when the solvent is water, the poly(carboxylic acid) polymer (A) may be reacted with the
5 polyvalent metal compound during application of the coating liquid, resulting in formation of non-uniform precipitates. Therefore, the solvent employed in the coating liquid is preferably a non-aqueous solvent, or a mixture of a non-aqueous solvent and water. As used herein, the term "non-
10 aqueous solvent" means a solvent other than water.

[0022]

The coating liquid containing the polyvalent metal compound (B) and a solvent may optionally contain, in addition the compound (B) and the solvent, an additive such
15 as a resin, a dispersant, a surfactant, a softening agent, a stabilizer, a film-forming agent, an anti-blocking agent, or a tackifier. Particularly, in order to improve the dispersibility and coating performance of the polyvalent metal compound, preferably, a resin which is soluble in the
20 solvent employed is added to the coating liquid. Preferred examples of the resin includes alkyd resin, melamine resin, acrylic resin, nitrocellulose, urethane resin, polyester resin, phenolic resin, amino resin, fluorocarbon resin, and epoxy resin. In the coating liquid, the ratio by weight of
25 the polyvalent metal compound (B) to the resin may be chosen accordingly. No particular limitation is imposed on the total amount of the polyvalent metal compound, resin and the

other additives in the coating liquid. However, from the viewpoint of coating performance, it is preferably within the range of 0.1-50 wt.%, more preferably within the range of 1-50 wt.%.

5 [0023]

When the coating liquid containing a poly(carboxylic acid) polymer (A) and a solvent, or the coating liquid containing a polyvalent metal compound (B) and a solvent is applied onto the support film, an adhesive may be applied in advance, in order to enhance adhesion between the support film and the layer (a) or the layer (b). When the layer (a) or the layer (b) is provided so as to come into contact with an additional layer other than the support film, or when the multi-layer film of the present invention is designed in consideration of contact of the multi-layer film with an additional layer other than the support film, in order to enhance adhesion between the additional layer and the layer (a) or the layer (b), a tackifier or an adhesive may be applied onto the outer surface of the layer (a) or the layer (b). No particular limitation is imposed on the adhesive to be employed, and specific examples of the adhesive include solvent-soluble resins employed for dry laminating, anchor coating, or primers, such as alkyd resin, melamine resin, acrylic resin, nitrocellulose, urethane resin, polyester resin, phenolic resin, amino resin, fluorocarbon resin, and epoxy resin.

[0024]

The multi-layer film of the present invention may be further laminated with an additional layer. An additional layer may be provided on the surface of the support film on which no layer has been laminated, or an additional layer may be provided on the surface of the layer (a) or the layer (b) on which no layer has been laminated. Specifically, the multi-layer film may have, for example, the following layer structure: (additional layer/support film/layer (a)/layer (b)), (additional layer/support film/additional layer/layer (a)/layer (b)), (support film/layer (a)/layer (b)/additional layer), (support film/layer (b)/layer (a)/additional layer), (support film/additional layer/layer (b)/layer (a)/layer (b)/additional layer), (additional layer/support film/layer (b)/layer (a)/layer (b)), or (support film/layer (b)/layer (a)/layer (b)/additional layer). The location of such an additional layer is not limited by these examples. The material of such an additional layer may be selected from among materials which can be employed for forming the support film. However, the additional layer material is not necessarily heat-shrinkable, so long as the resultant multi-layer film does not lose heat shrinkability. Examples of the additional layer material include layers of the aforementioned adhesives, tackifiers, and heat-sensitive tackifiers. One or more additional layers may be laminated on the multi-layer film in accordance with purposes; for example, the purposes of imparting, to the multi-layer film or sheet, strength, sealability (particularly for prevention

of gas invasion through an end portion of the film), easy-to-open property, good appearance, light shielding property, and moisture-barrier property. Lamination of the additional layer(s) is performed through coating of a layer material, or
5 through a known lamination method by use of a film-like or sheet-like layer material with or without use of an adhesive. Specific examples of the lamination method include a dry lamination method, a wet lamination method, and an extrusion lamination method.

10 **[0025]**

 The oxygen permeability of the thus-produced heat-shrinkable multi-layer film of the present invention having gas-barrier property as measured at 30°C and a relative humidity of 80% is preferably $500 \text{ cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less,
15 more preferably $100 \text{ cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less. After an object has been packaged with the multi-layer film of the present invention, and the resultant packaged product has been subjected to thermal shrinkage treatment, the oxygen permeability of the multi-layer film as measured at 30°C and
20 a relative humidity of 80% is preferably $500 \text{ cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less, more preferably $100 \text{ cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ or less. Thermal shrinkage treatment is performed by subjecting the packaged product to heat treatment at the temperature of preferably 50-130°C, more preferably 70-120°C, still more
25 preferably 80-100°C for preferably 1 second to 120 minutes, more preferably 1 second to 60 minutes, still more preferably 1 second to 30 minutes. Preferably, thermal shrinkage

treatment can be performed by use of, for example, hot water, vapor, steam, or hot air.

The multi-layer film of the present invention, which includes the heat-shrinkable support film coated with the layer (a) and the layer (b), exhibits sufficient heat shrinkability, which is attributed to the heat shrinkability of the support film while the layer (a) and the layer (b) composing the present invention do not have heat shrinkability in themselves. The oxygen permeability of the multi-layer film after thermal shrinkage as measured at 30°C and a relative humidity of 80% is preferably equal to or lower than that thereof before thermal shrinkage. The result indicates that even when the film is subjected to thermal shrinkage, at least gas-barrier property of the multi-layer film is not deteriorated.

[0026]

The heat-shrinkable multi-layer film of the present invention can be employed as a film, a sheet or a label; or can be formed into a packaging bag or a packaging container. Specific examples of the form of the packaging bag include a pillow packaging bag, a three-sides-sealed packaging bag, a four-sides-sealed packaging bag, and a gazette four-sides-sealed packaging bag. Specific examples of the form of the packaging container include a bottle, a tray, a cup, and a tube. For example, a container filled with an object to be packaged is entirely or partially covered with the multi-layer film of the present invention or with a packaging bag

formed of the multi-layer film, and the resultant product is subjected to thermal shrinkage treatment, so as to shrink the multi-layer film or the packaging bag and to bring the film or the bag into close contact with the container, whereby the resultant container ensures oxygen gas-barrier property.

When the materials of the multi-layer film are appropriately chosen, the multi-layer film can be provided with, for example, easy-to-open property, easy-to-tear property, shrinkability, applicability to microwave ovens, UV-shielding property, or good appearance. Particularly when the multi-layer film is employed as a label, the label is often used for covering only the body of a container. In this case, from the viewpoint of gas-barrier property, it is preferred that a tackifier (e.g., a heat-sensitive tackifier) is applied to the surface of the label that comes into contact with the container, and the label is caused to adhere to the container.

The multi-layer film of the present invention is suitable for use as a packaging material, packaging container, or vacuum heat-insulating material for articles which are readily impaired by, for example, oxygen, including foods, beverages, chemicals, drugs, and precision metallic parts such as electronic parts. In addition, the multi-layer film is suitable for use as a packaging material for articles which require long-term reliable gas-barrier property, and require treatment in high-temperature hot water (e.g., boiling or retort sterilization).

[0027]

Examples of food-related uses of the multi-layer film include packaging materials for convenience-store-related products, such as packed lunch, prepared food, cooked noodle, and *nabeyaki udon* (pot-boiled noodle); cover material of a container for pudding or fruit jelly; packaging materials for Chinese foods, and general prepared foods such as *tsukudani* (food boiled in soy sauce), pickles, and *nimame* (boiled beans); packaging materials for retort foods, Japanese confectionery, Western confectionery, processed marine products, processed meat products, fried foods, and fish-based paste products such as *kamaboko* (fish sausage) and *oden* (Japanese hotchpotch) ingredients; packaging materials for meats and fish-related products; and packaging materials for mushroom/vegetable-related products such as raw shiitake mushroom, *maitake* (*Grifola frondosa*), apple, banana, pumpkin, ginger, and *myoga* (Japanese ginger). The multi-layer film is also employed for single-, bundle-, or integrated-packaging of a beverage contained in a paper container, such as juice, milk, or a lactic acid bacteria beverage; or employed for single-, bundle-, or integrated-packaging of a beverage or food contained in a plastic container, such as a PET bottle beverage, juice, milk, a lactic acid bacteria beverage, or a pot noodle product.

Other examples of uses of the multi-layer film include packaging materials for paper products such as data sheets, photosensitive paper sheets, paper trays, and bag-in-box;

packaging materials for electric appliances, electric products, mechanical components, and building materials such as plywood, floor material, ceiling material, storm door, shutter, gate, fence, and stocker; packaging materials for furniture, office machinery, fiber, metallic coils, cutting boards, tableware, and household groceries such as aluminum foil; packaging materials for doughnut-shaped products such as pipe products, electric wire products, tube products, string products, band products, and electromagnetic wave shielding tube products; packaging materials for stationery such as notebooks, albums, and calendars; packaging materials for chemicals such as drugs, spray products (aerosol), and detergents; packaging materials for cosmetic and toiletry products such as hair-related products, soap, toothpaste, and wet tissue; packaging materials for audiovisual-related products such as CDs, cassette tapes, and videotapes; packaging materials for earthenware; packaging materials for sporting goods, fishing goods, building materials such as columns, precision parts, and gasoline tanks; and a material for integrated packaging of dry batteries. The multi-layer film can be employed as a label for the aforementioned various packaging materials. Particularly, the multi-layer film is suitable for use as a label of, for example, a paper container for storing juice, milk, a lactic acid bacteria beverage, or the like, or a PET container (bottle) for storing a beverage.

[0028]

[Examples]

The present invention will next be described in detail by way of Examples, which should not be construed as limiting the invention thereto. Evaluation methods and the Examples will be described below.

1. Evaluation methods for heat-shrinkable multi-layer film and the Examples

1.1 Measurement of percent thermal shrinkage

A film sample having dimensions of 10 cm × 10 cm was immersed in hot water at 90°C for 30 seconds, and then the percent thermal shrinkage of the film sample was calculated by use of the following formula:

$$\text{percent thermal shrinkage (\%)} = \{(L - L')/L\} \times 100$$

(wherein L and L' represent the lengths of the sample in a machine direction or a transverse direction before thermal shrinkage and after thermal shrinkage, respectively). The thus-determined percent thermal shrinkage in at least one direction of the film of the present invention is 3 to 90%.

1.2 Evaluation of appearance

A heat-shrinkable multi-layer film was subjected to 10% shrinkage in a transverse direction through the below-described procedure under dry heating or in hot water, and the appearance of the thus-shrunk film was evaluated on the basis of the below-described criteria.

1.2.1 10% Shrinkage under dry heating

A steel can (outer diameter: 53 mm, volume capacity: 250 cm³) was covered with a heat-shrinkable multi-layer film

which had been formed into a tubular shape such that when the film undergoes 10% shrinkage in a transverse direction, the film adheres to the steel can. Subsequently, the film-covered steel can was allowed to stand for one minute in a Geer oven whose temperature was regulated to 95°C, and was exposed to hot air, to thereby heat-shrink the film.

1.2.2 10% Shrinkage in hot water

A steel can (outer diameter: 53 mm, volume capacity: 250 cm³) was covered with a heat-shrinkable multi-layer film which had been formed into a tubular shape such that when the film undergoes 10% shrinkage in a transverse direction, the film adheres to the steel can. Subsequently, the film-covered steel can was immersed in hot water at 90°C for 30 seconds, to thereby heat-shrink the film. The appearance of the heat-shrinkable multi-layer film which had undergone 10% thermal shrinkage in a transverse direction was evaluated.

1.2.3 Evaluation criteria

Appearance of the thus-shrunk film was evaluated on the basis of the following criteria:

A: neither wrinkles nor slack is formed on the multi-layer film; cracking, breakage, and exfoliation do not occur in a coating layer; and the coating layer maintains transparency.

B: neither wrinkles nor slack is formed on the multi-layer film, but a coating layer exhibits impaired transparency.

C: wrinkles or slack is formed on the multi-layer film;

cracking, breakage, or exfoliation occurs in a coating layer;
and the coating layer exhibits impaired transparency.

[0029]

1.3 Oxygen permeability before and after shrinkage

Oxygen permeability of a heat-shrinkable multi-layer film was measured before and after thermal shrinkage of the film. Thermal shrinkage of the film was performed under dry heating and in hot water under the above-described 10% thermal shrinkage conditions.

The oxygen permeability of the film was measured by use of an oxygen permeability measuring apparatus (OXTRANTM 2/20, product of Modern Control) under the following conditions: temperature: 30°C and relative humidity (RH): 80%. The oxygen permeability was measured in accordance with JIS K-7126, B method (equal-pressure method) and the method specified by ASTM D3985-81. The thus-measured value is represented in units of $\text{cm}^3(\text{STP})/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$. As used herein, "(STP)" refers to standard conditions (0°C, 1 atm) for specifying the volume of oxygen.

[0030]

(Example 1)

A commercially available adhesive for anchor coating (AC) (DicdryTM LX747, product of Dainippon Ink and Chemicals, Inc., curing agent: KX75, solvent: ethyl acetate) was applied onto a heat-shrinkable polyester film (Space Clean S7542, product of Toyobo Co., Ltd., thickness: 45 μm , oxygen permeability: $600 \text{ cm}^3(\text{STP})/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$, percent shrinkage in

a machine direction: 5%, percent shrinkage in a transverse direction: 60% (as measured after immersion in hot water at 90°C for 30 seconds)) (hereinafter the film may be referred to as a "heat-shrinkable PET film") by use of a bar coater
5 (K303PROOFER, product of RK PRINT-COAT INSTRUMENT), followed by drying of the adhesive. The resultant coating layer was found to have a thickness of 1.0 μm . Separately, a poly(carboxylic acid) polymer (polyacrylic acid (PAA), Aron A-10H, product of Toagosei Co., Ltd., number average
10 molecular weight: 200,000, 25 wt.% aqueous solution) (hereinafter the polymer may be referred to as "PAA") was diluted with distilled water, to thereby prepare a 5 wt.% aqueous solution. The thus-prepared aqueous solution was applied onto the AC adhesive coating layer by use of the
15 aforementioned bar coater, followed by drying. The resultant coating layer was found to have a thickness of 0.3 μm .

Subsequently, in a manner similar to that described above, a zinc-oxide-fine-particles-containing resin (ZR133, product of Sumitomo Osaka Cement Co., Ltd., average particle
20 size: 0.02 μm , solid content: 33 wt.%, solvent for dispersion (toluene : MEK = 4 : 1)) was applied onto the above-dried PAA layer, followed by drying. The resultant coating layer was found to have a thickness of 1.0 μm . Thus, there was produced a multi-layer film having the following layer
25 structure: heat-shrinkable PET film (45 μm)/AC adhesive layer (1.0 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (1.0 μm). When the thus-produced multi-layer film was immersed in

hot water at 90°C for 30 seconds, the film exhibited a percent shrinkage of 5% in a machine direction and a percent shrinkage of 60% in a transverse direction.

[0031]

5 (Example 2)

In place of the AC adhesive, the zinc-oxide-fine-particles-containing resin (ZR133) employed in Example 1 was applied onto the aforementioned heat-shrinkable PET film in a manner similar to that of Example 1, followed by drying. The
10 resultant coating layer was found to have a thickness of 1.0 μm . Subsequently, the poly(carboxylic acid) polymer (PAA) employed in Example 1 was applied onto the coating layer, and then the aforementioned zinc-oxide-fine-particles-containing resin (ZR133) was applied onto the resultant PAA layer,
15 followed by drying, to thereby produce a multi-layer film having the following layer structure: heat-shrinkable PET film (45 μm)/ZnO-containing resin layer (1.0 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (1.0 μm). When the thus-produced multi-layer film was immersed in hot water at 90°C
20 for 30 seconds, the film exhibited a percent shrinkage of 5% in a machine direction and a percent shrinkage of 60% in a transverse direction.

[0032]

(Example 3)

25 The procedure of Example 1 was repeated, except that an adhesive for anchor coating was not employed, and the thickness of a zinc-oxide-fine-particles-containing resin

coating layer was changed. Specifically, PAA was applied onto the heat-shrinkable PET film, and the zinc-oxide-fine-particles-containing resin (ZR133) was applied onto the resultant PAA layer, followed by drying, to thereby produce a multi-layer film having the following layer structure: heat-shrinkable PET film (45 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (3.0 μm).

When the thus-produced multi-layer film was immersed in hot water at 90°C for 30 seconds, the film exhibited a percent shrinkage of 5% in a machine direction and a percent shrinkage of 60% in a transverse direction.

[0033]

(Example 4)

The procedure of Example 2 was repeated, except that a heat-shrinkable polyamide film (BONYL SC, product of Kohjin Co., Ltd., thickness: 15 μm , percent shrinkage in a machine direction: 20%, percent shrinkage in a transverse direction: 20% (as measured after immersion in hot water at 90°C for 30 seconds), oxygen permeability: 1,250 $\text{cm}^3(\text{STP})/(\text{m}^2\cdot\text{day}\cdot\text{MPa})$) (hereinafter the film may be referred to as a "heat-shrinkable ONY film") was employed as a support film. Specifically, the zinc-oxide-fine-particles-containing resin (ZR133) employed in Example 2 was applied onto the heat-shrinkable ONY film, followed by drying. The resultant coating layer was found to have a thickness of 1.0 μm . Subsequently, the PAA employed in Example 1 was applied onto the coating layer, and then the aforementioned zinc-oxide-

fine-particles-containing resin (ZR133) was applied onto the resultant PAA layer, followed by drying, to thereby produce a multi-layer film having the following layer structure: heat-shrinkable ONy film (15 μm)/ZnO-containing resin layer (1.0 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (1.0 μm). When the thus-produced multi-layer film was immersed in hot water at 90°C for 30 seconds, the film exhibited a percent shrinkage of 20% in a machine direction and a percent shrinkage of 20% in a transverse direction.

10 **[0034]**

(Example 5)

In a manner similar to that of Example 1, the adhesive for anchor coating (AC) and PAA were applied onto the heat-shrinkable PET film employed in Example 1, followed by drying. Subsequently, a mixture of an aqueous polyvinyl alcohol (PVA) solution (solid content: 10 wt.%) and an aqueous calcium acetate solution (containing calcium acetate produced by Wako Pure Chemical Industries, Ltd., concentration: 1.0 mol/kg) was applied onto the above-dried PAA layer in a manner similar to that described above, followed by drying. The resultant coating layer was found to have a thickness of 1.0 μm . Subsequently, a polyurethane (PU) resin (NEWLP SUPER RT medium, product of Toyo Ink Mfg. Co., Ltd., curing agent: VM Hardener XB, solvent: toluene, MEK) was applied onto the calcium-acetate-containing PVA coating layer, for the purpose of imparting water resistance. Thus, there was produced a multi-layer film having the following layer structure: heat-

shrinkable PET film (45 μm)/AC adhesive layer (1.0 μm)/PAA layer (0.3 μm)/calcium-acetate-containing PVA layer (1.0 μm)/PU resin layer (1.0 μm). When the thus-produced multi-layer film was immersed in hot water at 90°C for 30 seconds, the film exhibited a percent shrinkage of 5% in a machine direction and a percent shrinkage of 60% in a transverse direction.

[0035]

(Comparative Example 1)

10 The zinc-oxide-fine-particles-containing resin (ZR133) employed in Example 1 was applied onto the surface of the heat-shrinkable PET film employed in Example 1, followed by drying, to thereby produce a multi-layer film having the following layer structure: heat-shrinkable PET film (45
15 μm)/ZnO-containing resin layer (1.0 μm). When the thus-produced multi-layer film was immersed in hot water at 90°C for 30 seconds, the film exhibited a percent shrinkage of 5% in a machine direction and a percent shrinkage of 60% in a transverse direction.

20 Table 1 shows the results of evaluation of the multi-layer films produced in Examples 1 through 6 and Comparative Example 1.

[0036]

[Table 1]

Table 1

	Layer structure	Appearance evaluation		Oxygen permeability (cm ³ (STP)/m ² ·day·MPa)		
		After 10% shrinkage under dry heating	After 10% shrinkage in hot water	Before shrinkage	After 10% shrinkage under dry heating	After 10% shrinkage in hot water
Ref. Ex. 1	HSPET ^{*1} (45) μm	A	A	600	600	600
Ref. Ex. 2	HSony ^{*2} (15) μm	A	A	1250	1250	1250
Ex. 1	HSPET/AC ^{*3} /PAA ^{*4} /ZnO ^{*5} (45)/(1.0)/(0.3)/(1.0) μm	A	A	60	40	40
Ex. 2	HSPET/ZnOA/PAA/ZnOA (45)/(1.0)/(0.3)/(1.0) μm	A	A	40	10	10
Ex. 3	HSPET/PAA/ZnOA (45)/(0.3)/(3.0) μm	A	A	60	40	40
Ex. 4	HSony ^{*2} /ZnOA/PAA/ZnOA (15)/(1.0)/(0.3)/(1.0) μm	A	A	60	40	40
Ex. 5	HSPET/AC/PAA/CaAc ^{*6} /PU ^{*7} (45)/(1.0)/(0.3)/(1.0)/(1.0) μm	A	A	60	40	40
Comp. Ex. 1	HSPET/ZnO (45)/(1.0) μm	A	A	600	600	600

HSPET^{*1}: Heat-shrinkable polyester (Space Clean S7542,
product of Toyobo Co., Ltd., thickness: 45 μm)

HSony^{*2}: Heat-shrinkable polyamide (BONYL SC, product of
Kohjin Co., Ltd., thickness: 15 μm)

5 AC^{*3}: Adhesive for anchor coating (DicdryTM LX747, product of
Dainippon Ink and Chemicals, Inc., curing agent: KX75,
solvent: ethyl acetate)

PAA^{*4}: Polyacrylic acid (PAA) (Aron A-10H, product of
Toagosei Co., Ltd., number average molecular weight:
10 200,000, 25 wt.% aqueous solution diluted with distilled
water)

ZnO^{*5}: Zinc-oxide-fine-particles-containing resin (ZR133,
product of Sumitomo Osaka Cement Co., Ltd., average
particle size: 0.02 μm , solid content: 33 wt.%, solvent
15 for dispersion: toluene, MEK)

CaAc^{*6}: Polyvinyl alcohol containing calcium acetate (product
of Wako Pure Chemical Industries, Ltd.)

PU^{*7}: Polyurethane resin (NEWLP SUPER RT medium, product of
Toyo Ink Mfg. Co., Ltd., curing agent: VM Hardener XB,
20 solvent: toluene, MEK)

[0037]

2. Examples in which a heat-shrinkable multi-layer film is
employed as a label for a PET bottle, and is subjected to
evaluation

25 In the below-described Examples, tests were performed
with the intention of using a heat-shrinkable multi-layer
film for improving gas-barrier property of a container (e.g.,
a packaging container) containing an object. In the Examples,
in order to evaluate a heat-shrinkable multi-layer film of
30 the present invention, a packaging container was partially
covered with the multi-layer film; specifically, the multi-
layer film was employed as a label, and the body of a PET
bottle was partially covered with the film. In order to
evaluate gas-barrier property of the heat-shrinkable multi-

layer film, the film was combined with the packaging container by use of a tackifier.

2.1 Evaluation of appearance

A heat-shrinkable multi-layer film was combined with a PET bottle (volume capacity: 500 cm³), and the film was subjected to 10% shrinkage in a transverse direction through the below-described procedure under dry heating, followed by evaluation of the appearance of the thus-shrunk film.

2.1.1 10% Shrinkage under dry heating

A PET bottle was covered with a heat-shrinkable multi-layer film which had been formed into a tubular shape such that when the film undergoes 10% shrinkage in a transverse direction, the film adheres to the PET bottle. Subsequently, the film-covered PET bottle was allowed to stand for one minute in a Geer oven whose temperature was regulated to 95°C, and was exposed to hot air, to thereby heat-shrink the film. Thermal shrinkage was performed such that the surface area of a region of the PET bottle covered with the thus-shrunk multi-layer film accounts for about 80% of the entire surface area of the PET bottle. The appearance of the heat-shrinkable multi-layer film which had undergone 10% thermal shrinkage in a transverse direction and which had been combined with the PET bottle was evaluated.

2.1.2 Evaluation criteria

A: neither wrinkles nor slack is formed on the multi-layer film; the film adheres to the container; cracking, breakage, and exfoliation do not occur in a coating layer;

and the coating layer maintains transparency.

B: neither wrinkles nor slack is formed on the multi-layer film, and the film adheres to the container, but a coating layer exhibits impaired transparency.

5 C: wrinkles or slack is formed on the multi-layer film; the film does not adhere to the container; cracking, breakage, or exfoliation occurs in a coating layer; and the coating layer exhibits impaired transparency.

[0038]

10 2.2 Oxygen permeability after shrinkage

The oxygen permeability of the PET bottle which had been combined with the multi-layer film by use of a tackifier was measured through the following two methods: a method in which the bottle *per se* is subjected to measurement

15 (hereinafter this method will be referred to as "the package method"); and a method in which a portion of the bottle that is covered with the multi-layer film is cut out of the bottle, and the cut portion is subjected to measurement (hereinafter this method will be referred to as "the film method"). In
20 the package method, the bottle was fixated in a chamber (temperature: 30°C, relative humidity: 30%), nitrogen (relative humidity: 80%) was caused to flow through the bottle, and the amount of oxygen which permeates from the outside (air atmosphere) of the bottle into the bottle was
25 measured by use of the oxygen permeability measuring apparatus employed in 1.3 described above. The thus-measured value was multiplied by five, and the resultant value was

regarded as the value in the case where a space outside the bottle (inside the chamber) is filled with 100% oxygen; i.e., the oxygen permeability of the bottle. In the film method, the oxygen permeability was measured in a manner similar to that described above in 1.3. The oxygen permeability measured through the package method is represented in units of $\text{cm}^3(\text{STP})/(\text{bottle}\cdot\text{day}\cdot\text{MPa})$, whereas the oxygen permeability measured through the film method is represented in units of $\text{cm}^3(\text{STP})/(\text{m}^2\cdot\text{day}\cdot\text{MPa})$.

10 **[0039]**

(Example 6)

A heat-sensitive tackifier (Heat Magic DW4070, product of Toyo Ink Mfg. Co., Ltd.) was applied onto the surface of the ZnO-containing resin layer (opposite the surface of the heat-shrinkable PET film) of the multi-layer film of Example 2, such that the multi-layer film adheres to an object to be packaged during the course of thermal shrinkage of the film, followed by drying of the adhesive. The resultant heat-shrinkable multi-layer film (percent shrinkage in a machine direction: 5%, percent shrinkage in a transverse direction: 60% (as measured after immersion in hot water at 90°C for 30 seconds)) was formed into a cylindrical shape such that the adhesive layer faces inward. The body of a commercially available 500-mL PET bottle was covered with the thus-formed cylindrical multi-layer film. Subsequently, the film-covered PET bottle was allowed to stand for one minute in a Geer oven whose temperature was regulated to 95°C, and was exposed to

hot air, to thereby heat-shrink the film and combine the film with the bottle.

Thus, there was produced a packaging container having the following structure: heat-shrinkable PET film (45 μm)/ZnO-containing resin layer (1.0 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (1.0 μm)/tackifier layer (3.0 μm)/PET bottle (average thickness: 350 μm). The surface area of the multi-layer film combined with the PET bottle surface was found to account for 80% of the entire surface area of the PET bottle. The oxygen permeability of the PET bottle *per se* was found to be $2.0 \text{ cm}^3(\text{STP})/(\text{bottle}\cdot\text{day}\cdot\text{MPa})$ at 30°C and a relative humidity of 80%.

[0040]

(Example 7)

The heat-sensitive tackifier employed in Example 6 was applied onto the surface of the heat-shrinkable PET film of the multi-layer film of Example 2, such that the multi-layer film adheres to an object to be packaged during the course of thermal shrinkage of the film, followed by drying of the adhesive. The resultant heat-shrinkable multi-layer film (percent shrinkage in a machine direction: 5%, percent shrinkage in a transverse direction: 60% (as measured after immersion in hot water at 90°C for 30 seconds)) was formed into a cylindrical shape such that the tackifier layer faces inward. In a manner similar to that of Example 7, the body of a commercially available 500-mL PET bottle was covered with the thus-formed cylindrical multi-layer film.

Subsequently, the film-covered PET bottle was allowed to stand for one minute in a Geer oven whose temperature was regulated to 95°C, and was exposed to hot air, to thereby heat-shrink the film and combine the film with the bottle.

5 Thus, there was produced a packaging container having the following structure: ZnO-containing resin layer (1.0 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (1.0 μm)/heat-shrinkable PET film (45 μm)/tackifier layer (3.0 μm)/PET bottle (average thickness: 350 μm). The surface area of the
10 multi-layer film combined with the PET bottle surface was found to account for 80% of the entire surface area of the PET bottle.

[0041]

(Comparative Example 2)

15 The heat-sensitive tackifier employed in Example 6 was applied onto the heat-shrinkable PET film employed in Example 1, followed by drying. The resultant heat-shrinkable multi-layer film (percent shrinkage in a machine direction: 5%, percent shrinkage in a transverse direction: 60% (as measured
20 after immersion in hot water at 90°C for 30 seconds)) was formed into a cylindrical shape such that the adhesive layer faces inward. In a manner similar to that of Example 6, the body of a commercially available 500-mL PET bottle was covered with the thus-formed cylindrical multi-layer film.
25 Subsequently, the film-covered PET bottle was allowed to stand for one minute in a Geer oven whose temperature was regulated to 95°C, and was exposed to hot air, to thereby

heat-shrink the film and combine the film with the bottle.
Thus, there was produced a packaging container having the
following structure: heat-shrinkable PET film (45
 μm)/tackifier layer (3.0 μm)/PET bottle (average thickness:
5 350 μm). The surface area of the heat-shrinkable PET film
combined with the PET bottle surface was found to account for
80% of the entire surface area of the PET bottle.

Table 2 shows the results of evaluation of the multi-
layer films produced in Examples 6 and 7 and Comparative
10 Example 2.

[0042]

[Table 2]

Table 2

	Layer structure	Appearance evaluation	Oxygen permeability	
			Package method (cm ³ (STP)/bottle·day·MPa)	Film method (cm ³ (STP)/m ² ·day·MPa)
Ref. Ex. 3	PETb ^{*1} (average: 350 μm)	-	2.0	50
Ex. 6	HSPET ^{*2} /ZnO ^{*3} /PAA ^{*4} /ZnO/Ad ^{*5} /PETb (45)/(1.0)/(0.3)/(1.0)/(3.0)/(350)	A	0.4	5.0
Ex. 7	ZnO/PAA/ZnO/HSPET/Ad/PETb (1.0)/(0.3)/(1.0)/(45)/(3.0)/(350)	A	0.2	3.0
Comp. Ex. 2	HSPET/Ad/PETb (45)/(3.0)/(350)	A	2.0	50

PETb^{*1}: Commercially available PET bottle (volume capacity: 500 cm³)

HSPET^{*2}: Heat-shrinkable polyester (Space Clean S7542, product of Toyobo Co., Ltd., thickness: 45 μm)

ZnO^{*3}: Zinc-oxide-fine-particles-containing resin (ZR133, product of Sumitomo Osaka Cement Co., Ltd., average particle size: 0.02 μm, solid content: 33 wt.%, solvent for dispersion: toluene, MEK)

PAA^{*4}: Polyacrylic acid (PAA) (Aron A-10H, product of Toagosei Co., Ltd., number average molecular weight: 200,000, 25 wt.% aqueous solution diluted with water)

Ad^{*5}: Heat-sensitive tackifier (Heat Magic DW4070, product of Toyo Ink Mfg. Co., Ltd.)

[0043]

3. Examples in which a heat-shrinkable multi-layer film is employed for entirely covering an object, and is subjected to evaluation

5 In the below-described Examples, in order to evaluate a heat-shrinkable multi-layer film, an object is entirely covered with the film. Specifically, a sausage product was packaged with the heat-shrinkable multi-layer film, and then the film was subjected to evaluation.

10 (Example 8)

 Through dry lamination, a heat-shrinkable polyethylene film (Polyset UM, Product of Kohjin Co., Ltd., thickness: 35 μm , percent shrinkage in a machine direction: 15%, percent shrinkage in a transverse direction: 18% (as measured after
15 immersion in hot water at 90°C for 30 seconds)) was attached, via a polyurethane adhesive (Takelac A620, product of Mitsui Takeda Chemicals, Inc., curing agent: Takenate A65, solvent: ethyl acetate), onto the surface of the ZnO-containing resin layer of the heat-shrinkable multi-layer film produced in
20 Example 4, to thereby produce a multi-layer film having the following layer structure: heat-shrinkable ONy film (15 μm)/ZnO-containing resin layer (1.0 μm)/PAA layer (0.3 μm)/ZnO-containing resin layer (1.0 μm)/adhesive layer (2 μm)/heat-shrinkable polyethylene film (35 μm). A sausage
25 product was packaged with the thus-produced multi-layer film by use of a bag-forming and packaging machine (ONPACK-6600AII, product of Orihiro Co., Ltd.). The resultant packaged

product was immersed in hot water at 90°C for 10 minutes, to thereby perform thermal shrinkage and simultaneously thermal sterilization. After completion of thermal shrinkage, substantially no change in the shape of the sausage product was observed, and the packaged product exhibited good appearance, with the film being sufficiently tightly fitted to the sausage product. No breakage was observed in the thus-shrunk film. The oxygen permeability of the thus-shrunk film was found to be $1.0 \text{ cm}^3(\text{STP})/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ at 30°C and a relative humidity of 80%.

[0044]

(Comparative Example 3)

By means of dry lamination, the heat-shrinkable polyethylene film employed in Example 8 was attached, via the adhesive employed in Example 8, onto one surface of the heat-shrinkable ONy film employed in Example 4, to thereby produce a multi-layer film having the following layer structure: heat-shrinkable ONy film (15 μm)/adhesive layer (2 μm)/heat-shrinkable polyethylene film (35 μm). A sausage product was packaged with the thus-produced multi-layer film by use of the bag-forming and packaging machine described above in Example 8. The sausage-packaged product was immersed in hot water at 90°C for 10 minutes, to thereby perform thermal shrinkage and simultaneously thermal sterilization. After completion of thermal shrinkage, substantially no change in the shape of the sausage product was observed, and the packaged product exhibited good appearance, with the film

being sufficiently tightly fitted to the sausage product. No breakage was observed in the thus-shrunk film. The oxygen permeability of the thus-shrunk film was found to be 1,200 $\text{cm}^3(\text{STP})/(\text{m}^2 \cdot \text{day} \cdot \text{MPa})$ at 30°C and a relative humidity of 80%.

5 **[0045]**

[Effects of the Invention]

The present invention provides a heat-shrinkable multi-layer film which is a packaging material exhibiting gas-barrier property, and which is employed for shrink packaging
10 of an object without use of a container or a cover material, or for shrink packaging of the entirety of a container.

[Document Name] Abstract

[Abstract]

[Object] To provide a multi-layer film exhibiting heat shrinkability without impairing gas-barrier property.

5 **[Constitution]** A heat-shrinkable multi-layer film comprising a heat-shrinkable support film (base film) and, on at least one surface of the base film, at least one layer structure including a layer (a) formed of a poly(carboxylic acid) polymer (A) and a layer (b) formed of a polyvalent metal
10 compound (B), the layers (a) and (b) being in contact with each other, and the multi-layer film exhibiting a percent thermal shrinkage of 5 to 90%.

[Selected Figure of Drawings] None